

# SYLLABUS

## CHEMISTRY (043) (THEORY) CLASS XII (2021-22) TERM-II

| Time : 2 Hours |   | Marks : 35     |       |
|----------------|---|----------------|-------|
| S.No.          | Unit                                    | No. of Periods | Marks |
| 1              | Electrochemistry                        | 7              |       |
| 2              | Chemical Kinetics                       | 5              | 13    |
| 3              | Surface Chemistry                       | 5              |       |
| 4              | d- and f-Block Elements                 | 7              |       |
| 5              | Coordination Compounds                  | 8              | 9     |
| 6              | Aldehydes, Ketones and Carboxylic Acids | 10             |       |
| 7              | Amines                                  | 7              | 13    |
| Total          |   | 49             | 35    |

**Electrochemistry :** Redox reactions, EMF of a cell, standard electrode potential, Nernst equation and its application to chemical cells, Relation between Gibbs energy change and EMF of a cell, conductance in electrolytic solutions, specific and molar conductivity, variations of conductivity with concentration, Kohlrausch's Law, electrolysis.

**Chemical Kinetics :** Rate of a reaction (Average and instantaneous), factors affecting rate of reaction : concentration, temperature, catalyst, order and molecularity of a reaction, rate law and specific rate constant, integrated rate equations and half-life (only for zero and first order reactions).

**Surface Chemistry :** Adsorption - physisorption and chemisorption, factors affecting adsorption of gases on solids, colloidal state : distinction between true solutions, colloids and suspension; lyophilic, lyophobic, multi-molecular and macromolecular colloids; properties of colloids; Tyndall effect, Brownian movement, electrophoresis, coagulation.

**d- and f-Block Elements :** General introduction, electronic configuration, occurrence and characteristics of transition metals, general trends in properties of the first row transition metals – metallic character, ionisation enthalpy, oxidation states, ionic radii, colour, catalytic property, magnetic properties, interstitial compounds, alloy formation.

**Lanthanoids -** Electronic configuration, oxidation states and lanthanoid contraction and its consequences.

**Coordination Compounds :** Coordination compounds - Introduction, ligands, coordination number, colour, magnetic properties and shapes, IUPAC nomenclature of mononuclear coordination compounds. Bonding, Werner's theory, VBT and CFT.

**Aldehydes, Ketones and Carboxylic Acids :** Aldehydes and Ketones : Nomenclature, nature of carbonyl group, methods of preparation, physical and chemical properties, mechanism of nucleophilic addition, reactivity of alpha hydrogen in aldehydes, uses.

**Carboxylic Acids :** Nomenclature, acidic nature, methods of preparation, physical and chemical properties; uses.

**Amines :** Nomenclature, classification, structure, methods of preparation, physical and chemical properties, uses, identification of primary, secondary and tertiary amines.